

# Interpretation of the Cu K-edge EXAFS spectra of Cu<sub>3</sub>N using *ab initio* molecular dynamics

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Copper nitride (Cu<sub>3</sub>N) has a cubic anti-ReO<sub>3</sub>-type structure composed of NCu<sub>6</sub> octahedra joined by corners. The dynamic properties of its lattice are determined by the anisotropic thermal vibrations of copper atoms, having stronger amplitude perpendicular to linear N-Cu-N atomic chains. As a result, Cu<sub>3</sub>N has low decomposition temperature of about 300-470°C. Also its thermal displacement parameters suggest some degree of positional disorder at the Cu site, which can be responsible for the structural instability of Cu<sub>3</sub>N under pressure. X-ray absorption spectroscopy is ideally suited to probe in-situ local atomic structure of Cu<sub>3</sub>N as a function of external conditions [1, 2]. However, the analysis of total Cu K-edge EXAFS spectrum of Cu<sub>3</sub>N is a challenging task due to outer shells overlap and multiple-scattering contributions.

Our preliminary attempts to use classical molecular dynamics to interpret EXAFS of Cu<sub>3</sub>N, based on the approach suggested in [3], failed to describe accurately the lattice dynamics and, particularly, correlation effects already in the second coordination shell of copper. Therefore, in this study, we performed *ab initio* molecular dynamics (AIMD) simulations of Cu<sub>3</sub>N in the temperature range from 300 K to 700 K in order to elucidate the details of lattice dynamics and anharmonicity of the Cu-N and Cu-Cu bonding. The AIMD simulations were done based on Kohn-Sham density functional theory (DFT) by the CP2K code using “Piz Daint” supercomputer at the Swiss National Supercomputing Centre.

To validate the obtained theoretical results, we performed a direct comparison with the Cu K-edge EXAFS using the MD-EXAFS approach developed in [3]. We demonstrate the efficiency of the MD-EXAFS approach in combination with AIMD method for theory validation and temperature dependent structural properties description in Cu<sub>3</sub>N. The advantages and limitations of this approach are discussed.

[1] J. Timoshenko, A. Anspoks, A. Kalinko, A. Kuzmin, *Acta Mater.* 129 (2017) 61-71.

[2] A. Kuzmin, A. Kalinko, A. Anspoks, J. Timoshenko, R. Kalendarev, *Latvian J. Phys. Tech. Sci.* 53 (2016) 31-37.

[3] A. Kuzmin and R. A. Evarestov, *J. Phys.: Condens. Matter* 21 (2009) 055401.

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